Contents lists available at [ScienceDirect](http://www.sciencedirect.com/science/journal/00404039)

Tetrahedron Letters

journal homepage: www.elsevier.com/locate/tetlet

Convergent total synthesis of the racemic HIF-1 inhibitor laurenditerpenol

Michael E. Jung *, G-Yoon J. Im

Department of Chemistry and Biochemistry, University of California, Los Angeles, CA 90095-1569, United States

article info

ABSTRACT

Article history: Received 11 April 2008 Revised 20 May 2008 Accepted 23 May 2008 Available online 21 June 2008 The convergent total synthesis of the HIF-1 inhibitor laurenditerpenol 1a is reported. The key step is the Julia olefination–reduction process between the two components, the sulfone 4 (prepared from the dimethylfuran-maleic anhydride Diels–Alder adduct) and the aldehyde 3 (prepared from 3-methylcyclohexenone).

© 2008 Published by Elsevier Ltd.

Tetrahedroi

Laurenditerpenol 1, a secondary metabolite isolated from the Jamaican red alga Laurencia intricata in 2004 by Zhou and Nagel,¹ was shown to inhibit activation of the hypoxia inducible factor-1 (HIF-1) under hypoxia using a T47D human breast tumor cellbased reporter assay with an IC_{50} of 0.4 μ M. Upregulation of HIF-1 in late-stage tumors activates oxygen sensitive genes leading to angiogenesis and results in the survival and further proliferation of the tumor.² Inhibition of HIF-1 represents an alternate strategy in late-stage cancer therapy.³ In the original isolation paper, the stereochemistry at C6 and C7 was not determined, nor was the relative stereochemistry of the oxabicycloheptane unit versus the cyclohexenol. Very recently, Avery and coworkers published a synthesis of laurenditerpenol and many of its diastereomers, which proved the relative and absolute structure to be that shown in 1a (Scheme 1). 4 This publication prompts us to report our synthesis of laurenditerpenol 1a and its stereoisomers in racemic form by a completely different route to the one used by Avery. Our convergent synthetic route allowed for the easy installation of the C6 and C7 stereocenters and a late-stage coupling of the two major subunits.

The retrosynthetic strategy involved a disconnection at C8 and C9 to give two key intermediates, the aldehyde 3 and the sulfone 4, which would be coupled via a modified Julia olefination procedure to give the alkene 2 and thence 1a (Scheme 2). The aldehyde 3 would be prepared from the lactone 5 while the sulfone 4 would

Scheme 1.

The synthesis of the aldehyde 3 ([Scheme 3\)](#page-1-0) began with the reduction of the known ester enone 8 (prepared by alkylation of the 3-methylcyclohexenone) 5 to give a diastereomeric mixture of the allylic alcohols 9 in 8[6](#page-2-0)% yield.⁶ Saponification to the hydroxy acid and lactonization using DCC yielded the two lactones 10 and 11 as a separable 2:1 mixture. Methylation of the enolate of the major lactone 10 occurred from the less hindered face to generate in 85% yield the methyl lactone 12, which on DIBAL reduction gave the lactol 13. Similarly, methylation of the trans lactone 11 occurred cis to the allylic hydrogen to generate the methylated product 14 in modest yield.

The oxanorbornane sulfone 4 was synthesized starting from the known exo Diels–Alder adduct of 2,5-dimethylfuran and maleic anhydride 7, which had been originally prepared by Diels and Alder in 1928^7 1928^7 [\(Scheme 4](#page-1-0)). Hydrogenation gave the anhydride,

Corresponding author. Tel.: +1 310 825 7954; fax: +1 310 206 3722. E-mail address: jung@chem.ucla.edu (M. E. Jung).

^{0040-4039/\$ -} see front matter © 2008 Published by Elsevier Ltd. doi:10.1016/j.tetlet.2008.05.116

which was opened to the known carboxylic acid ester 15^{7c} via methanolysis. Treatment of 15 with sodium methoxide in methanol took advantage of the difference in pK_a for protons α to an ester versus those α to a carboxylate salt, a principle we had used before in the synthesis of cyclobut-A analogues, 8 to give cleanly the epimerized ester 16. Selective reduction of the carboxylic acid with borane furnished in 89% yield the alcohol, which was converted to the tosylate 17 in 83% yield. Reduction of both the ester and the tosylate with lithium aluminum hydride unveiled the third methyl group to afford the trimethyl oxanorbornane alcohol 18 in 87% yield. The alcohol was converted to the sulfide 20 (96%) via treatment with 2-mercaptobenzothiazole 19 under Mitsunobu conditions, for example, diisopropyl azodicarboxylate (DIAD) and triphenylphosphine, which was then oxidized with m-CPBA to yield the desired sulfone 4 in 95% yield.

Unfortunately, the key Julia–Kocienski olefination⁹ between the lactol 13 and the sulfone 4 was unsuccessful presumably due to the fact that the lactol exists predominately in the cyclic isomer with little or none of the hydroxyaldehyde being present. The required free aldehyde 21 was therefore prepared via the 5-step protecting group manipulation starting from the lactone 12, that is, reduction to the diol (85%), monoprotection as the pivalate (89%), silylation

and reductive removal of the pivalate (86%), and final oxidation to give the aldehyde 21 in 78% yield (Scheme 5).^{[10](#page-2-0)} The modified Julia olefination of the protected hydroxyaldehyde 21 and the sulfone 4 yielded the desired set of racemic trans products 22E and 23E, along with the racemic cis products 22Z and 23Z in 88% yield as a 1:1 mixture of the Z and E isomers.¹¹ The E isomers could be separated from the Z isomers by flash chromatography. We planned to carry out selective reduction of the disubstituted alkene in the presence of the cyclic trisubstituted alkene via reduction with diimide¹² (prepared by protonation of dipotassium azodicarboxylate by acetic acid in a solution of pyridine and methanol). The hindered Z isomers-22Z and 23Z-did not react with diimide and the starting alkenes were recovered unchanged. However, the mixture of the E isomers, 22E and 23E, underwent diimide reduction to afford, in addition to 15% starting material, a 1:1 mixture of the desired reduced products 24 and 25 in 80% yield, as observed in the NMR. Unfortunately, the unreduced olefin and the desired alkane diastereomers were difficult to separate via column chromatography, and only the diastereomer 25 was isolated in small quantities. Desilylation of the TBS ether of 25 with TBAF afforded the laurenditerpenol diastereomer 1b in 72% yield. TBAF deprotection of the TBS ethers of the mixture of the alkenes and alkanes gave both laurenditerpenol diastereomers 1a and 1b, which were partially separated by careful chromatography. The proton, and especially the carbon, NMRs of these isolated diastereomers matched those reported by Avery and coworkers (which were numbered 1c and ent-1g in their paper).¹

We have also prepared several analogues of laurenditerpenol and are submitting them for biological evaluation as potential

Scheme 6.

HIF-1 inhibitors (Scheme 6). Thus hydrolysis of the silyl ethers of the four alkene stereoisomers—22E/23E and 22Z/23Z—with TBAF in THF afforded the dienols 26E/27E and 26Z/27Z, which were separated by careful flash column chromatography.¹³ The results of the biological assays of these compounds will be reported in due course.

Finally, we were able to prepare the correct (S)-enantiomer of the allylic alcohol 9 using the Corey–Bakshi–Shibata (CBS) reduction¹⁴ of the cyclohexenone **8**, even though **8** has a stereocenter α to the ketone. This reduction gave a diastereomeric mixture of the allylic alcohols 9 in 60% yield and 85% de.¹⁵ We are currently investigating a chiral synthesis of the oxabicyclo[2.2.1]heptane sulfone 4 (via a chemical or enzymatic resolution of the anhydride 7) with the idea of carrying out a more efficient chiral synthesis of the correct enantiomer. Our results will be reported in due course.

In summary, we have described a convergent synthesis of racemic laurenditerpenol 1a and its diastereomer 1b, in 13 steps from 3-methyl-2-cyclohexen-1-one. The methylated oxanorbornane fragment 4 was successfully synthesized from 2,5-dimethylfuran and coupled to the aldehyde 21 via a high-yielding modified Julia olefination procedure. A more detailed account of the synthesis, along with efforts toward the synthesis of analogues, will be forthcoming.

Acknowledgments

We thank the National Science Foundation (CHE 0614591) for generous support of this work.

References and notes

- 1. Mohammed, K. A.; Hossain, C. F.; Zhang, L.; Bruick, R. K.; Zhou, Y.-D.; Nagle, D. G. J. Nat. Prod. 2004, 67, 2002–2007.
- 2. Schumaker, P. T. Crit. Care Med. 2005, 33, S423–S425.
3. Giaccia A: Siim B. G.: Johnson B. S. Nat. Rev. Drug E.
- 3. Giaccia, A.; Siim, B. G.; Johnson, R. S. Nat. Rev. Drug Disc. **2003**, 2, 803–811.
4. Chittibovina, A. G.: Kumar, G. M.: Carvalho, P. B.: Liu, Y.: Zhou, Y.-D.: Nagle 4. Chittiboyina, A. G.; Kumar, G. M.; Carvalho, P. B.; Liu, Y.; Zhou, Y.-D.; Nagle, D.
- G.; Avery, M. A. J. Med. Chem. 2007, 50, 6299–6302.
- 5. Podraza, K. F.; Bassfield, R. L. J. Org. Chem. **1989**, 54, 5919–5922.
6. Podraza, K. F. US Patent 5 114 493, 1992.
- 6. Podraza, K. F. US Patent 5,114,493, 1992.
7. (a) Diels O: Alder K.: Naujoks E. Ber. De
- 7. (a) Diels, O.; Alder, K.; Naujoks, E. Ber. Deutsch. Chem. Gesell. Abteilung B 1929, 62B, 554–562; (b) Brickwood, D. J.; Ollis, W. D.; Stephanatou, J. S.; Stoddart, J. F. J. Chem. Soc., Perkin Trans. 1 1978, 1398–1414; (c) Imagawa, T.; Matsuura, K.; Murai, N.; Akiyama, T.; Kawanisi, M. Bull. Chem. Soc. Jpn. 1983, 56, 3020–3022; (d) Akiyama, T.; Fujii, T.; Ishiwari, H.; Imagawa, T.; Kawanisi, M. Tetrahedron Lett. 1978, 19, 2165-2166.
- 8. Jung, M. E.; Sledeski, A. W. J. Chem. Soc., Chem. Commun. 1993, 589–591.
- 9. (a) Baudin, J. B.; Hareau, G.; Julia, S. A.; Ruel, O. Bull. Soc. Chim. Fr. 1993, 130, 336–357; (b) Baudin, J. B.; Hareau, G.; Julia, S. A.; Lorne, R.; Ruel, O. Bull. Soc. Chim. Fr. 1993, 130, 856–878; (c) Bellingham, R.; Jarowicki, K.; Kocienski, P.; Martin, V. Synthesis 1996, 285–296.
- 10. (a) Nielsen, T. E.; Le Quement, S.; Juhl, M.; Tanner, D. Tetrahedron 2005, 61, 8013–8024; (b) Tanner, D.; Tedenborg, L.; Somfai, P. Acta Chem. Scand. 1997, 51, 1217–1223.
- 11. For a discussion of the various effects on the stereochemistry of the Julia– Kocienski reaction, see: Blakemore, P. R.; Cole, W. J.; Kocienski, P. J.; Morley, A. Synlett 1998, 26–28.
- 12. (a) Kluge, A. F.; Untch, K. G.; Fried, J. H. J. Am. Chem. Soc. 1972, 94, 9256–9258; (b) Pasto, D. J.; Taylor, R. T. Org. React. 1991, 40, 91-155.
- 13. The structures of $26E$ and $27E$ are secure since they could be correlated with the reduction products 1ab by NMR data. However, the relative configurations of 26EZ could be inverted since we have not yet been to convert them into the reduced products which would have allowed a direct correlation.
- 14. (a) Corey, E. J.; Bakshi, R. K.; Shibata, S. J. Am. Chem. Soc. 1987, 109, 5551–5553; (b) Corey, E. J.; Bakshi, R. K.; Shibata, S.; Chen, C. P.; Singh, V. K. J. Am. Chem. Soc. 1987, 109, 7925–7926.
- 15. (a) For other preparations of 9, see: Omoto, M.; Yukawa, C.; Saijo, T. Jpn. Kokai Tokkyo Koho (2004) JP 2004269463; Chem. Abstr. 2004, 141, 314460.; For the synthesis of the racemic cis-lactone, see: (b) Bartlett, P. A.; Pizzo, C. F. J. Org. Chem. 1981, 46, 3896–3900; (c) Guth, H. Helv. Chim. Acta 1996, 79, 1559– 1571.